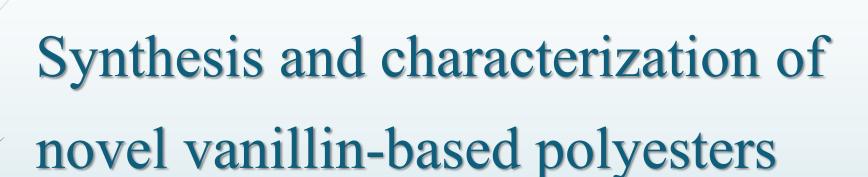


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Background

Most commodity plastics are fossil-based

Issues: * Progressive depletion of fossil resources

* Fluctuating oil price

Solution

✓ Biomass valorization for new environment friendly polymers

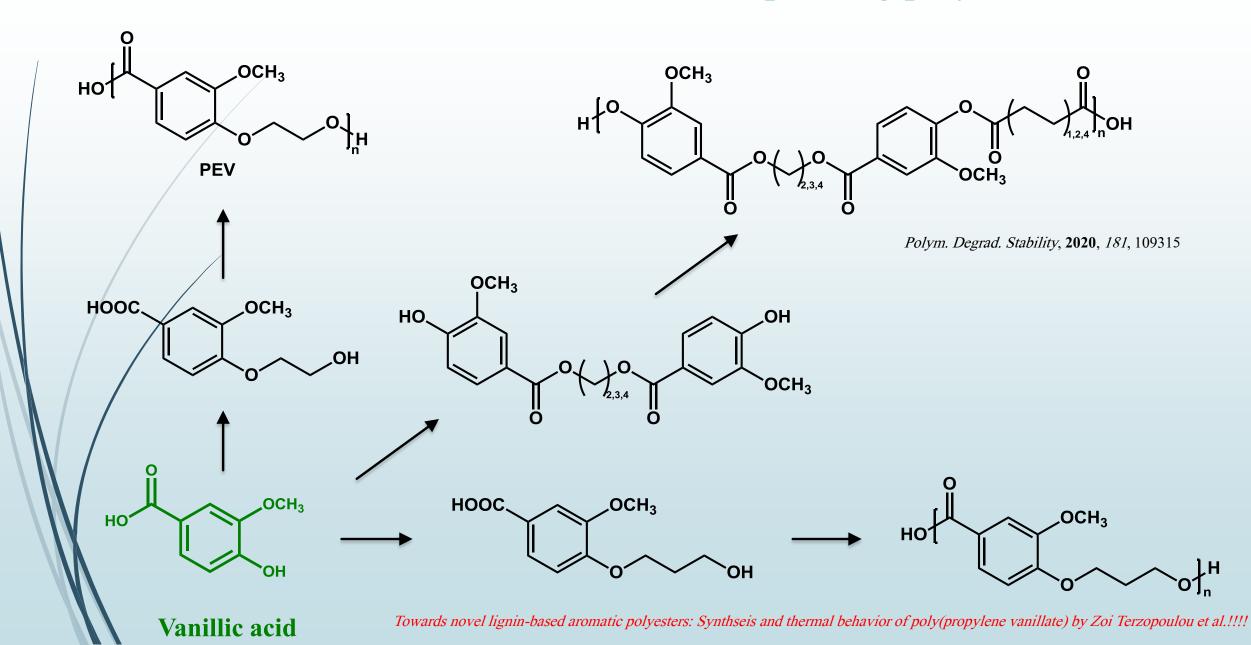
Lignin

Renewable, non-edible, low-cost by-product of the paper industry

Valorization in the production of novel bio-based monomers

- Vanillin produced from lignin is already commercially available.
- Vanillic acid, an aromatic hydroxy-acid, can be readily produced from vanillin by oxidation.

Bio-based monomers and corresponding polyesters



Poly(ethylene vanillate), PEV

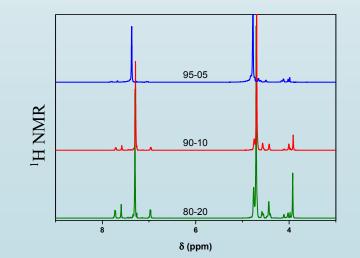
structurally similar to

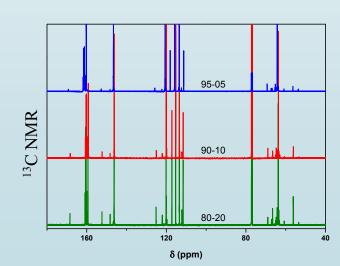
- \blacksquare T_g 83 °C, T_m 261 °C (comparable to PET T_g=80 °C and T_m=252 °C).
- $ightharpoonup \Delta H_{\rm m}^{\circ}$ 166 ± 16 J/g, equilibrium melting temperature ($T_{\rm m}^{0}$) 301.4 °C.
- Slow crystallization attributed to poor nucleation density.
- ► Thermal stability: 1% mass loss at 327 °C, maximum decomposition rate at ~420 °C.
- Thermal degradation mechanism (Py-GC/MS):
 - heterolytic scission characteristic of ester bonds, and
 - extensive C-O and C-C homolytic cleavage attributed to the ether linkage.
- Mechanical properties (nanoindentation): hardness close to PET.

Copolymers with ethylene furanoate



- Successful copolymerization demonstrated by NMR, probably block structure.
- ightharpoonup T_gs around 80 °C.
- Lower thermal stability compared to poly(ethylene furanoate).





Conclusions

- Vanillic acid is an interesting bio-based monomer for the synthesis of sustainable thermoplastics.
- Poly(ethylene vanillate) is a promising ether-containing polyester, with thermal properties comparable to PET.
- Poly(propylene vanillate) has exhibited remarkable thermal properties.
- Copolymers with ethylene furanoate are under investigation.

